

# **ECE606: Solid State Devices**

## **Lecture 1: Introduction**

Muhammad Ashraful Alam  
alam@purdue.edu

# Outline

- 1) Course information**
- 2) Current flow in semiconductors
- 3) Types of material systems
- 4) Classification of crystals

**Reference:** Vol. 6, Ch. 1 (pages 1-10)

# Course Information

## Books

- Advanced Semiconductor Fundamentals (QM, SM, Transport)
- Semiconductor Device Fundamentals (Diode, Bipolar, MOSFET)

## Website

<http://cobweb.ecn.purdue.edu/~ee606/>

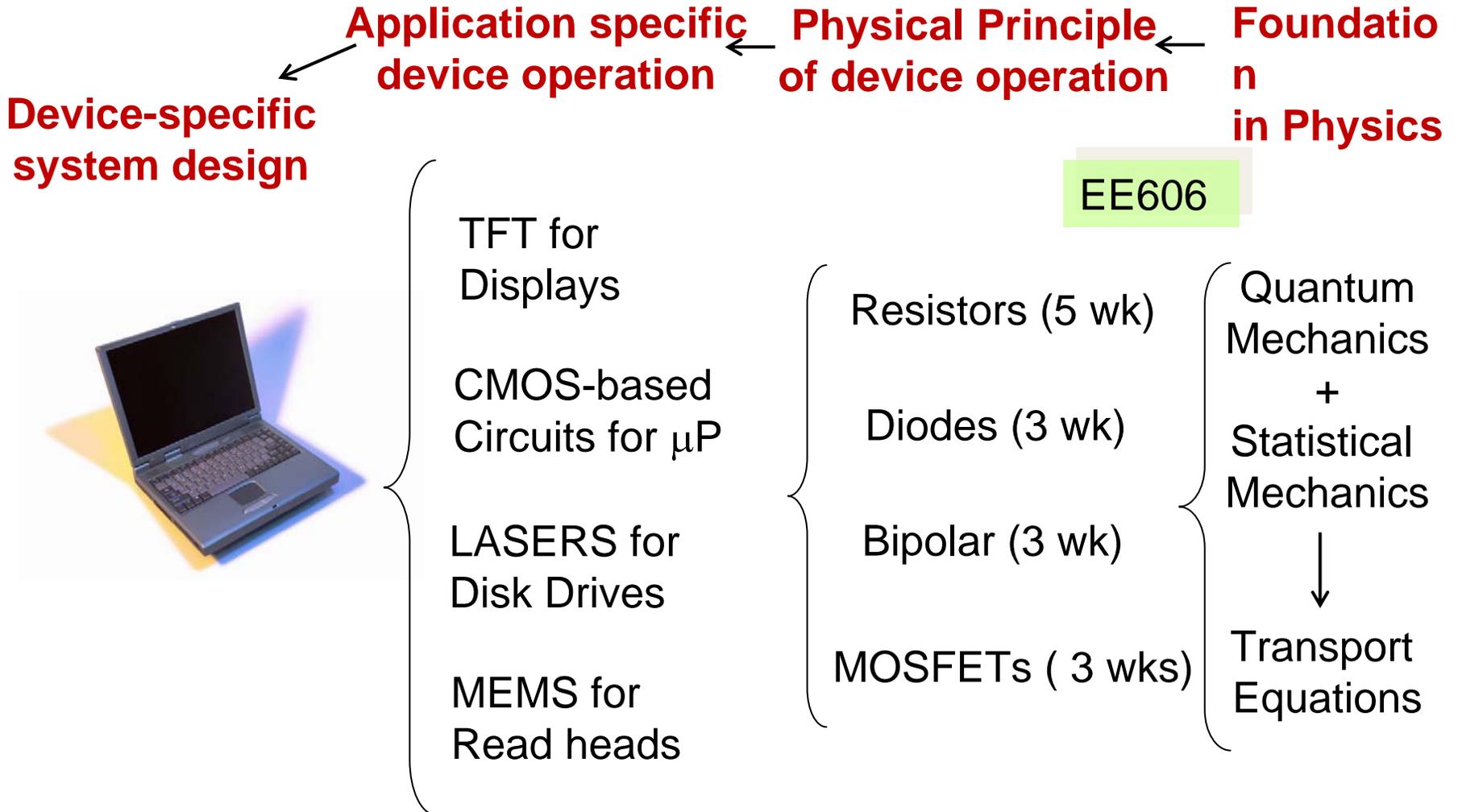
**Office hours** 8:20-9:15 MWF @ EE 320

## HW/Exams

HW (9 HW, 4 will be graded; solutions will be provided)

3 exams (~5 weeks apart)

# Outline of the Course



# Relation to Other MN-Area Courses

Device-specific system design

EE 695F:  
RF Design

EE 695E:  
Opto-system

BME 695A:  
Bio-system Design

Application specific device operation

EE 654:  
Advanced Semi Dev.

EE 612:  
VLSI Devices

EE 520:  
Bio-Systems

Physical Principle of device Operation

EE 606:  
Basic Semi Dev.

EE 604  
EM, Magnetics

Foundation

EE 656:  
Semi-Transport

EE 659:  
Quantum Transport

PHYS 570B :  
Bio-physics

Finite Element,  
Molecular Dynamics,  
Monte Carlo

Characterization

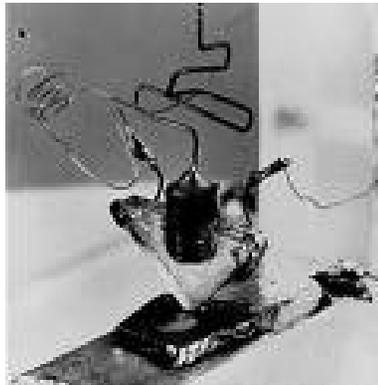
# Grand Challenges in Electronics

Vacuum Tubes



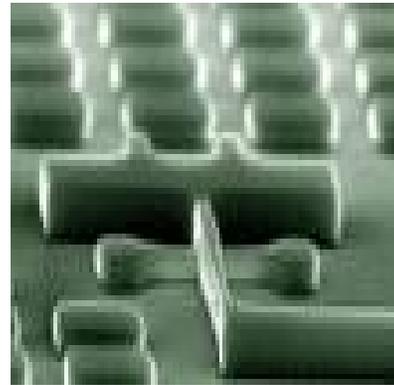
1906-1950s

Bipolar



1947-1980s

MOSFET



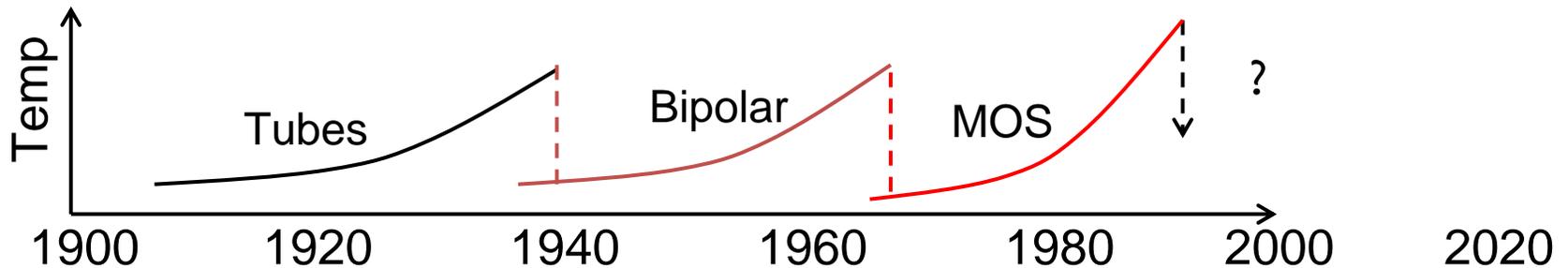
1960-until now

Now ??

Spintronics

Bio Sensors

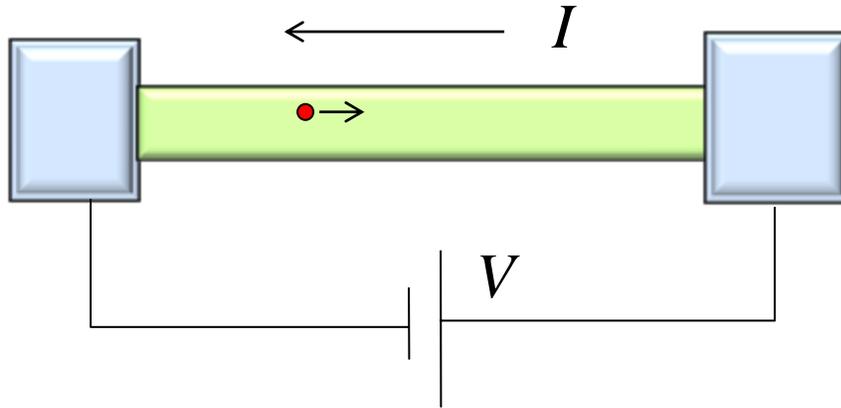
Displays ....



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# Current Flow Through Semiconductors (5 weeks)



$$I = G \times V$$

$$= q \times n \times v \times A$$

Carrier  
Density

velocity

Depends on chemical composition,  
crystal structure, temperature, doping, etc.

## Quantum Mechanics + Equilibrium Statistical Mechanics

⇒ Encapsulated into concepts of effective masses  
and occupation factors (Ch. 1-4)

## Transport with scattering, non-equilibrium Statistical Mechanics

⇒ Encapsulated into drift-diffusion equation with  
recombination-generation (Ch. 5 & 6)

# Computing Carrier-Density and Velocity

## Atomic composition

- *number of electrons per atom*

## Arrangement of atoms

- *not all electrons are available for conduction*

## For Periodic Arrays

- *simplification for computation*
  - ⇒ Concept of Unit Cells
  - ⇒ Simple 3-D Unit Cells

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# Elemental and Compound Semiconductors

↓ II	↓ III	↓ IV	↓ V	↓ VI
4 Be	5 B	6 C	7 N	8 O
12 Mg	13 <b>Al</b>	14 <b>Si</b>	15 <b>P</b>	16 S
30 <b>Zn</b>	31 <b>Ga</b>	32 Ge	33 <b>As</b>	34 <b>Se</b>
48 Cd	49 In	50 Sn	51 Sb	52 Te
80 Hg	81 Tl	82 Pb	83 Bi	84 Po

**Elemental** (e.g., Si, C)

**Compound**

IV-IV: Si-Ge, Si-C

III-V: InP, GaAs,  
(In<sub>x</sub>Ga<sub>1-x</sub>)(As<sub>y</sub>P<sub>1-y</sub>)

y)

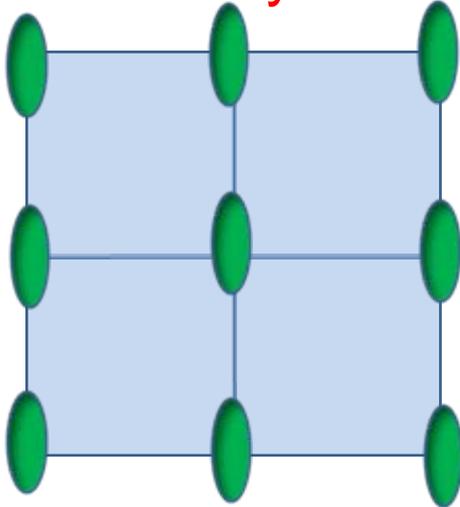
II-VI: CdTe

IV-VI: PbS

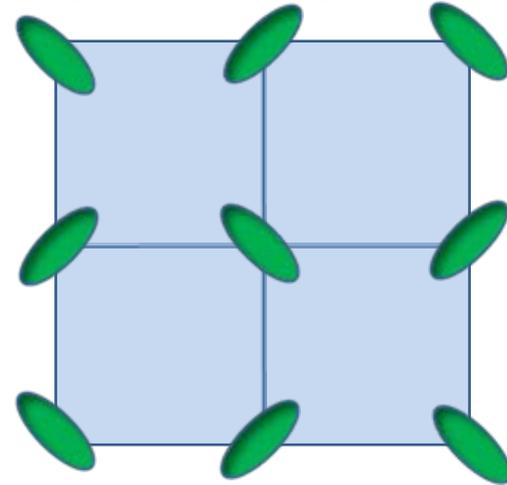
Not all combinations possible:  
lattice mismatch, room temp. instability, etc. are concerns

# Arrangement of Atoms: orientation vs. position

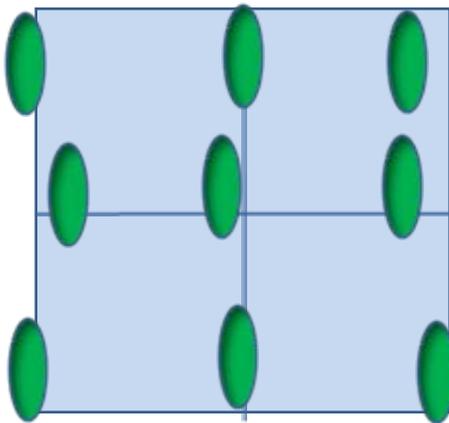
solid crystals



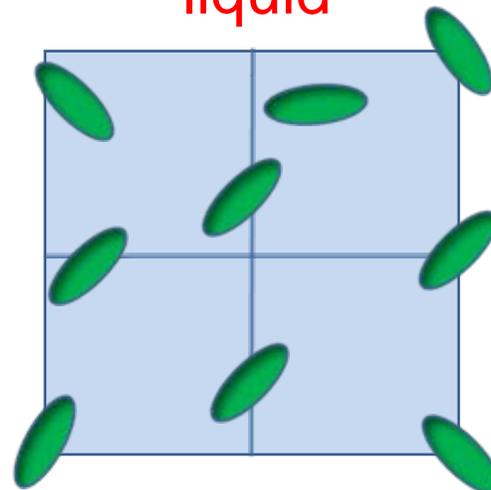
plastic crystals



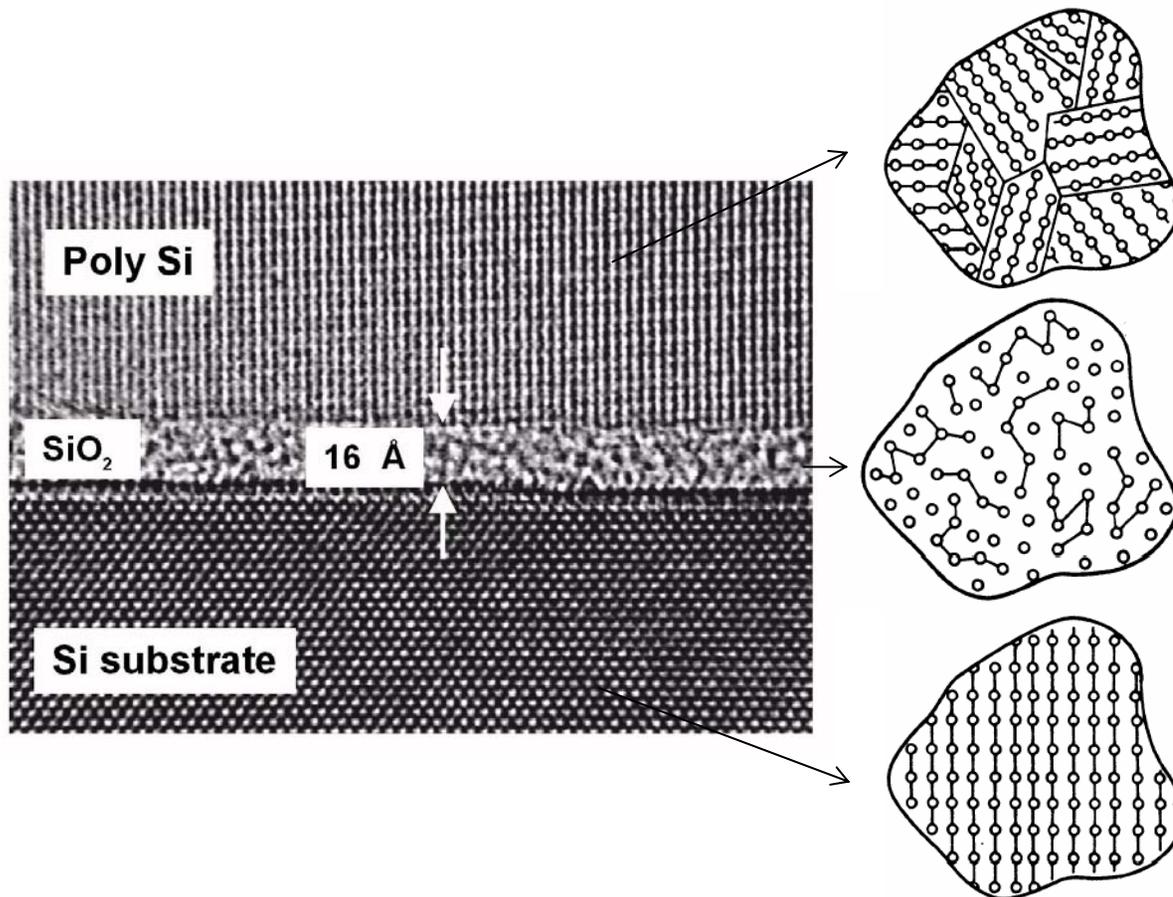
liquid crystals



liquid



# Arrangement of Atoms



Poly-crystalline  
Thin Film  
Transistors

Amorphous  
Oxides  
*Why ?*

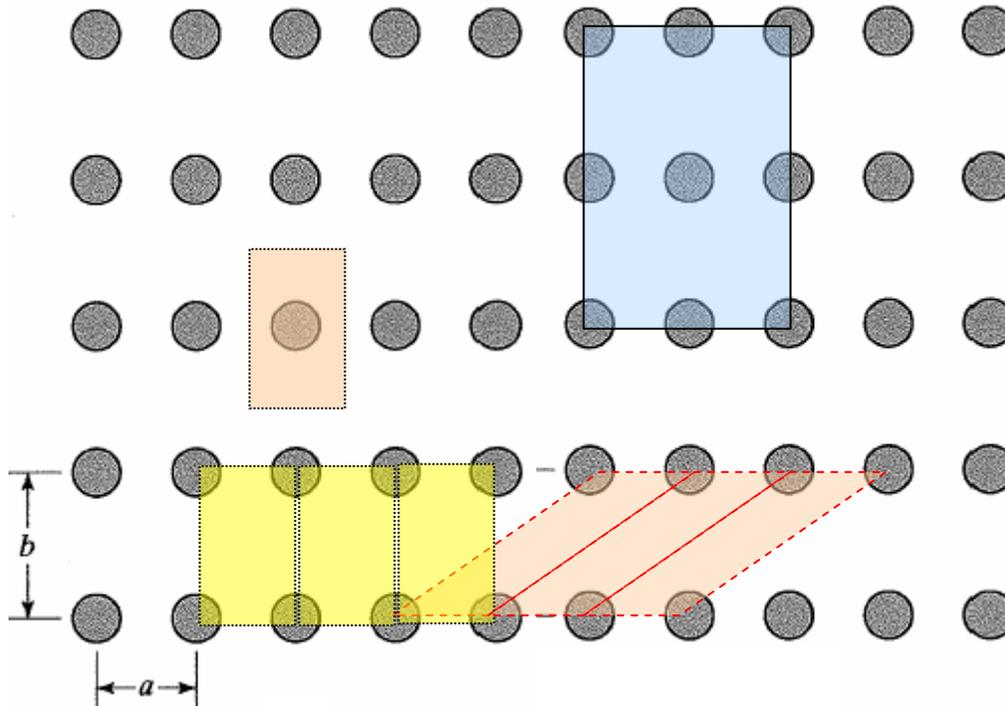
Crystalline  
*Definition ?*

- Quantitative definition: Correlation spectrum and diffraction pattern
- Modern solid state devices use all forms these forms of materials

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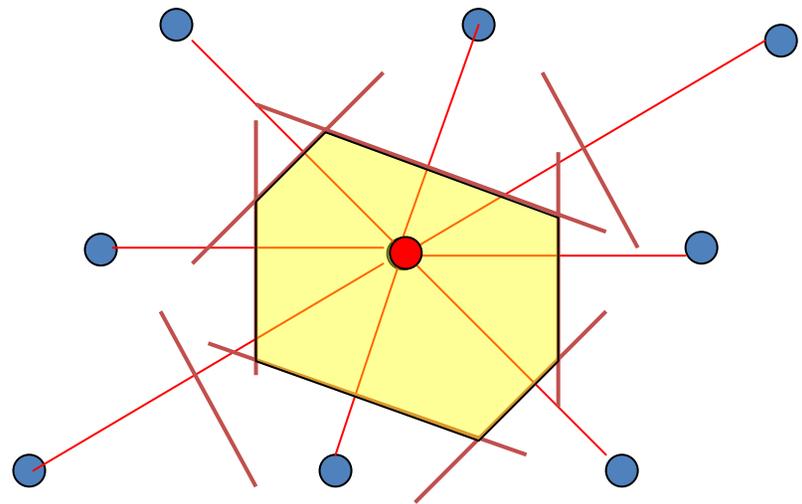
# Unit cell of a Periodic Lattice



- Unit cells are *not* unique
- Unit cells can be Primitive or Non-primitive
- Property of one defines the property of the solid

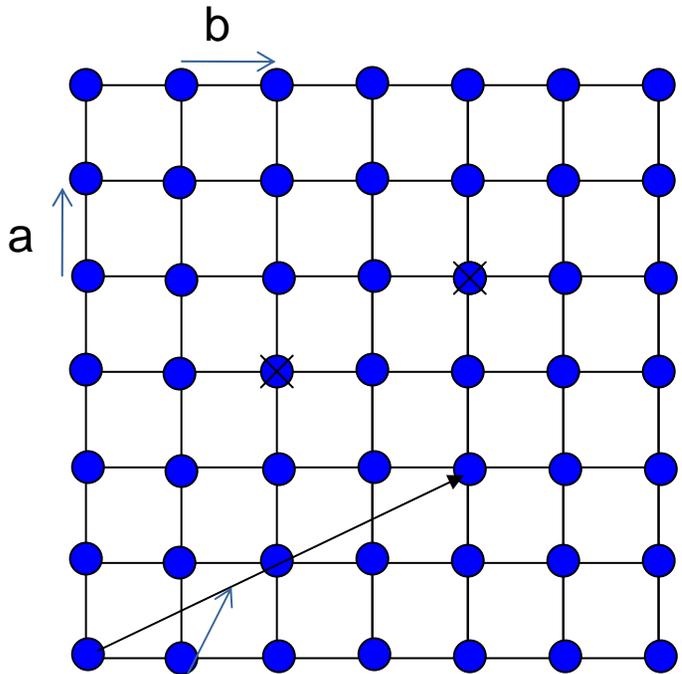
# Wigner-Seitz Primitive Cell

- Choose a reference atom
- Connect to all its neighbors by straight lines
- Draw lines (in 2D) or planes (in 3D) normal to and at the midpoints of lines drawn in step 2
- Smallest volume enclosed is the Wigner-Seitz primitive cell

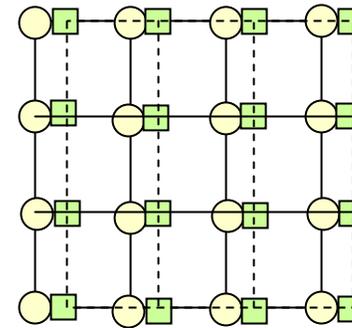


# Geometry of Lattice Points

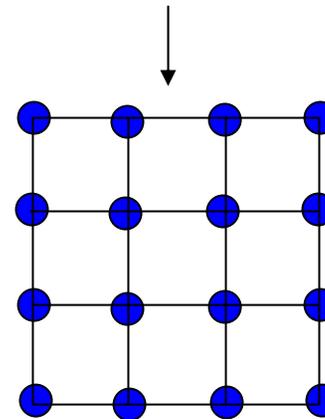
In a Bravais lattice, every point has the same environment as every other point (same number of neighbors, next neighbors,



$$\vec{R} = h\vec{a} + k\vec{b}$$



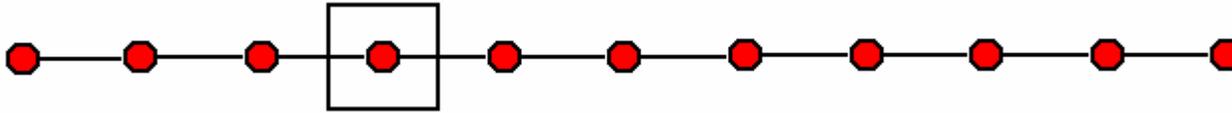
Non-Bravais lattice



Bravais lattice  
with a basis

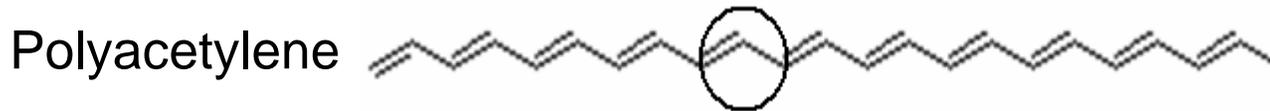


# Unit Cells in One-dimensional Crystals

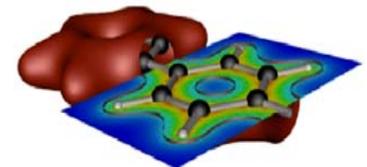
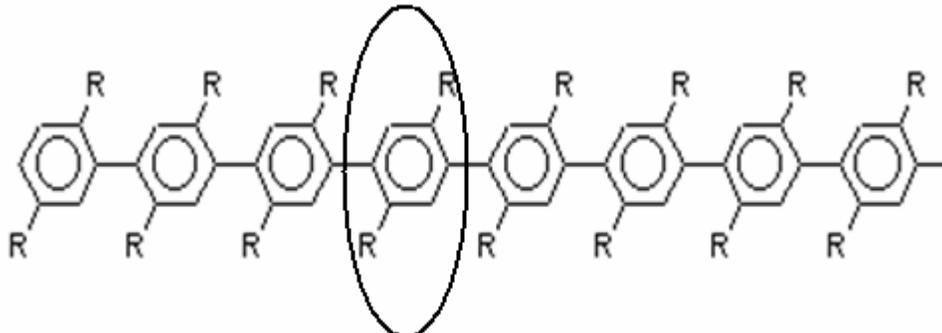


No system truly 1-D, but ....

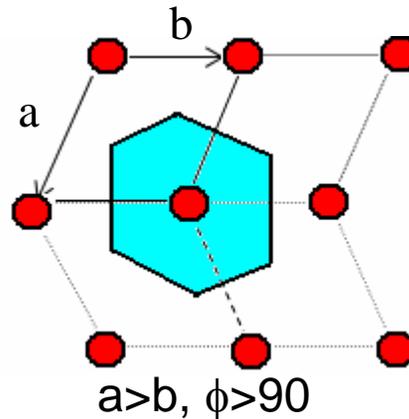
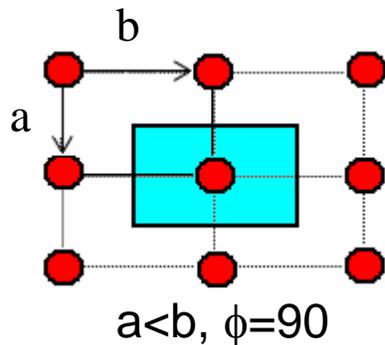
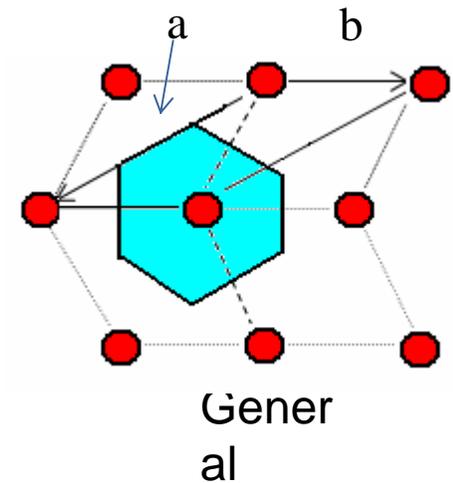
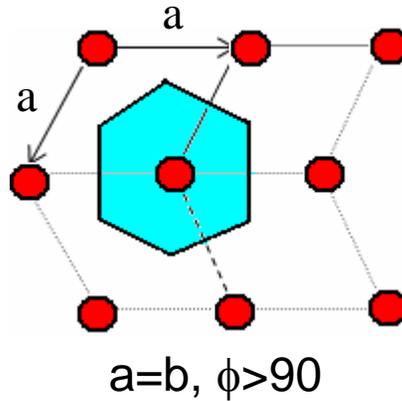
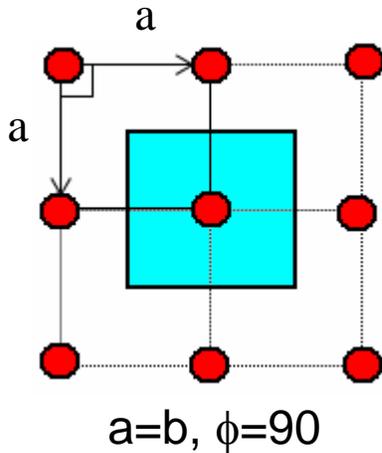
- 1D properties dominate behavior in some material (e.g., polymers)
- Can be solved analytically, many properties have 2D/3D analogs



PPP

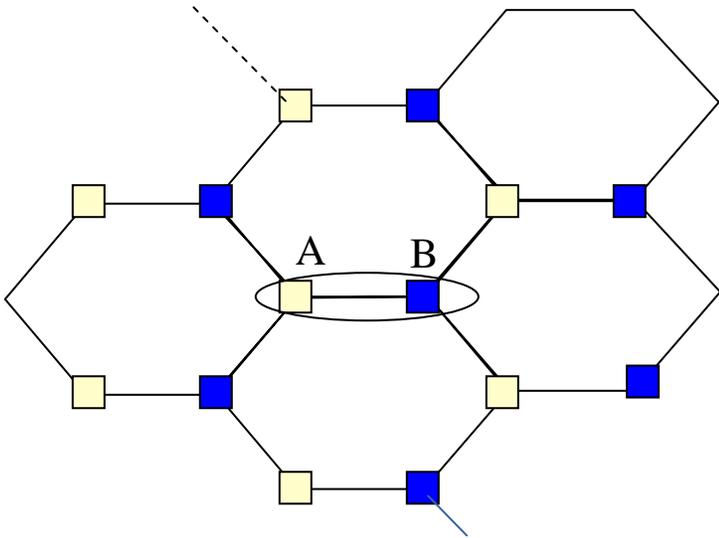


# Periodic Lattice in 2D (5-types)



- Many systems have 2D lattice (e.g., Graphene, Wigner crystals, ...)

# Not a Bravais Lattice ...



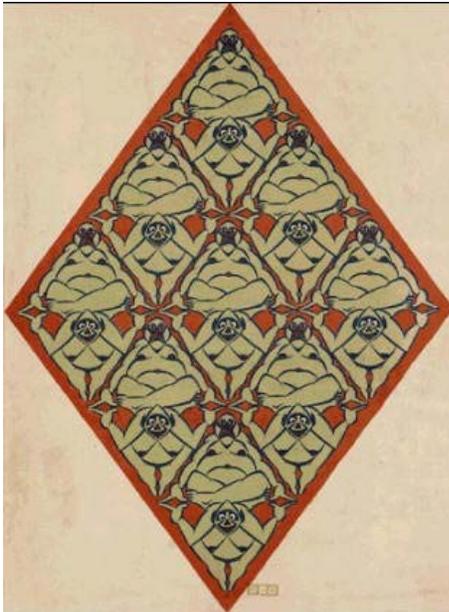
**A and B do not have  
identical environments**

This is a Graphene sheet which has recently been isolated from Graphite by adhesive tape stamping.

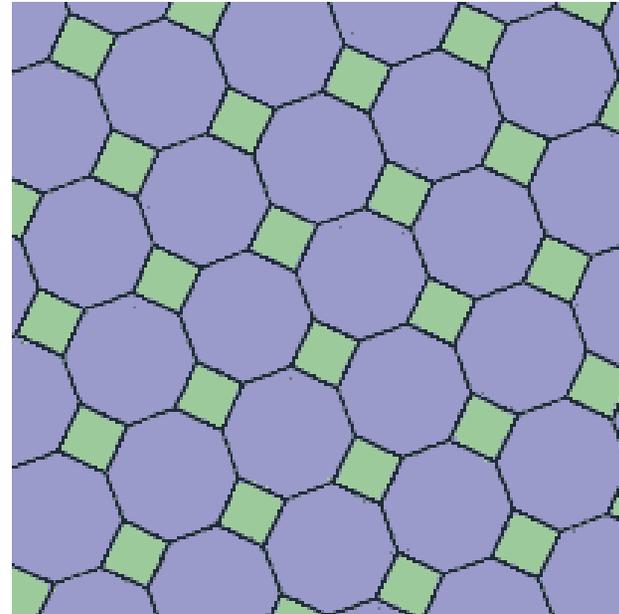
Ref. Novoselov, Geim, et al.  
Nature, 438, 197, 2005.

# Not a Bravais Lattice, but ...

Escher Tiling



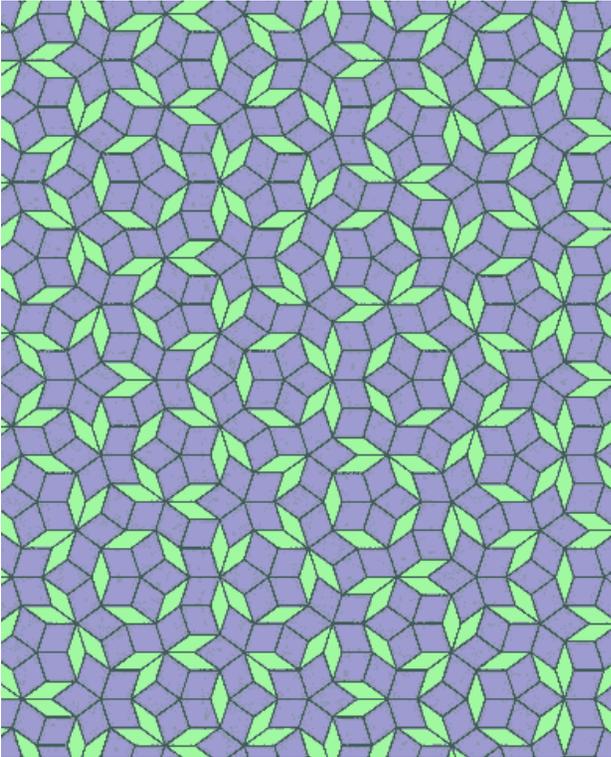
Kepler Tiling



....but these can be converted into Bravais lattice

# Not a Bravais Lattice and ...

## Penrose Tiles

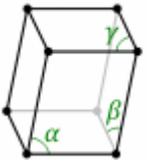
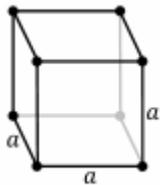
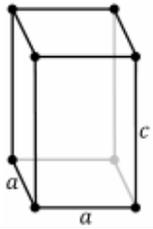
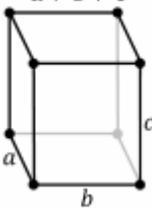
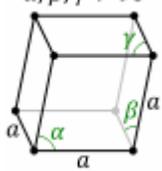
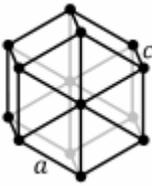
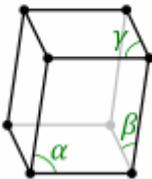
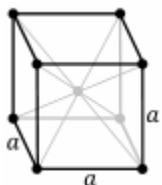
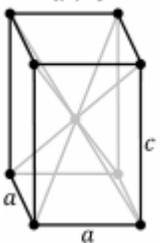
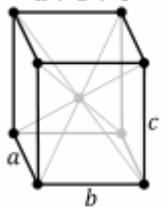
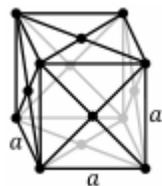
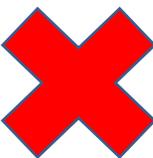
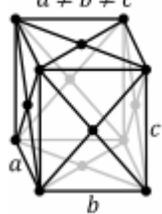
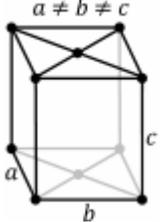
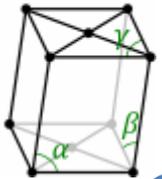


## Ancient Tiles



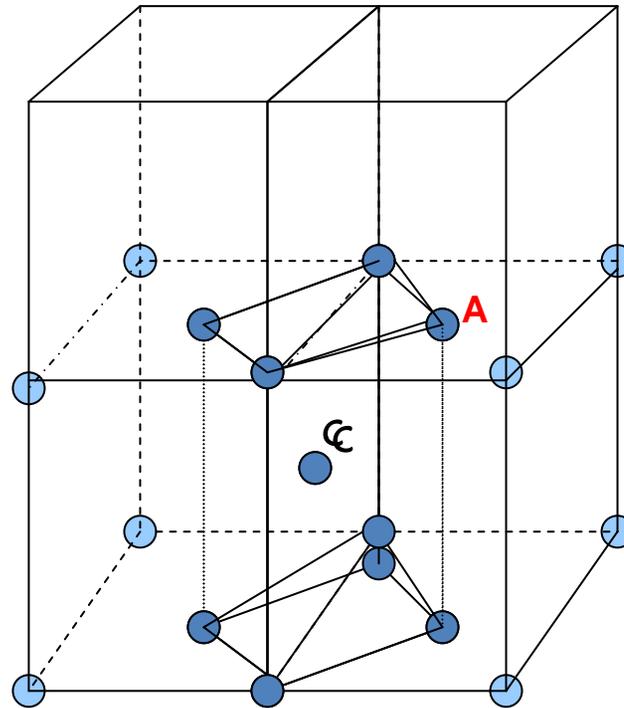
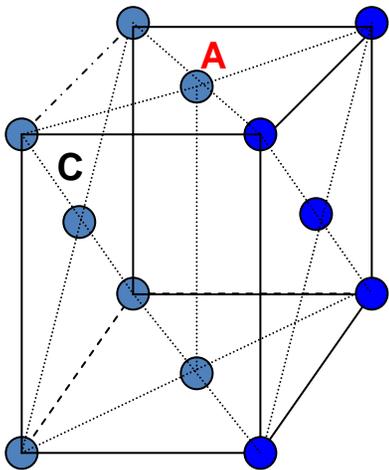
... these **CANNOT** be transformed to Bravais lattice  
ex. Aluminum-Manganese compounds, non-sticky coats

# Bravais lattice in 3D (14-types)

	Triclinic	Cubic	Tetragonal	Orthorobmic	Rhombohedral	Hexagonal	Monoclinic
P	$\alpha, \beta, \gamma \neq 90^\circ$ 		$a \neq c$ 	$a \neq b \neq c$ 	$\alpha, \beta, \gamma \neq 90^\circ$ 	$a \neq c$ 	$\alpha \neq 90^\circ$ $\beta, \gamma = 90^\circ$ 
I			$a \neq c$ 	$a \neq b \neq c$ 			
F				$a \neq b \neq c$ 			
C				$a \neq b \neq c$ 			$\alpha \neq 90^\circ$ $\beta, \gamma = 90^\circ$ 

# Duplicated Bravais Lattice

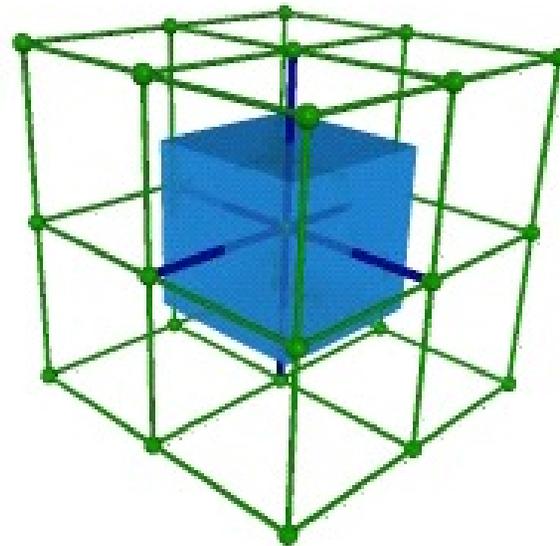
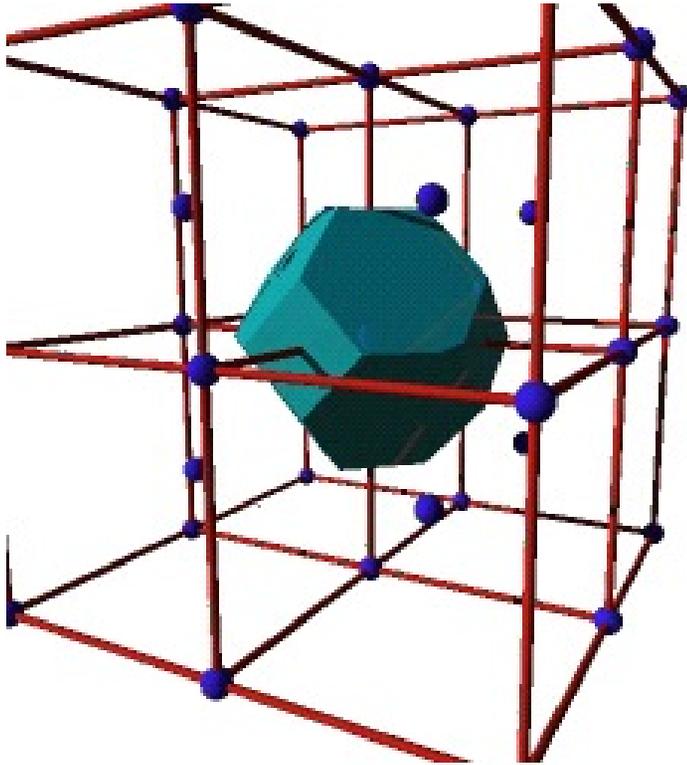
Unlucky Frankenheim (1842)!



A

Tetragonal BC = Tetragonal FC

# Wigner-Seitz cells for BCC and Cubic Lattices



Laura Malcolm, 1997

# Conclusions

1. Transport in semiconductors depends on carrier density ( $n$ ) and carrier velocity ( $v$ ). In order to find these quantities, we need to understand the chemical composition and atomic arrangements.
2. There is a wide variety of semiconductor materials (IV, III-V, II-IV, etc.) with different chemical compositions.
3. Crystalline materials can be built by repeating the basic building blocks. This simplifies the quantum solution of electronic states, which will allow us to compute ( $n$ ) and ( $v$ ) for these systems easily.